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NaCl{aq) + AgNQiaq) - AgCl(s) + NaN03(aq) H2S0iaq) + 2NaOH(aq) - N¾S04(aq) + 2H20(l) N¾S03(aq) + 2HCl(aq) - 2NaCl(aq) + Hp(l) + SOz(g) A compound burns in the presence of oxygen, producing energy in the form of heat and light. The combustion of organic compounds produces carbon dioxide and water.

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Summary. In this inquiry-based experiment, the students design and carry out a sequence of chemical reactions for the separation and identification of iron, silver, and zinc ions in water. Two parallel series of tests will be carried out, one using a known sample containing all three metal cations and the other using unknown samples containing one, two, or all three cations.

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